

SYNTHESIS OF A NITROGEN-FREE DERIVATIVE
OF JERVINE

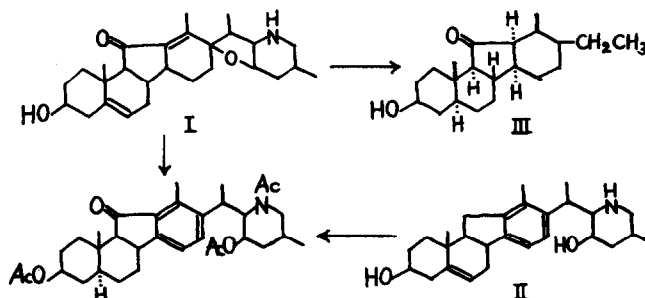
Hiroshi Mitsuhashi and Yuzuru Shimizu

Faculty of Pharmaceutical Sciences, Hokkaido University

Sapporo, Japan

(Received 10 October 1961; in revised form 13 November 1961)

The veratrum alkaloids, jervine (I) and veratramine (II) possess the C-nor-D-homosteroid skeleton as determined by Wintersteiner and co-workers.¹ Recently K. Tsuda, S. Okuda and H. Kataoka have determined the absolute configuration of the methyl group in the piperidine residue,² but the correlation of jervine-derived products with derivatives of conventional steroids has been lacking. We have now succeeded in obtaining the degradation product (III) of Fried and Klingsberg³ in the course of our studies of C-nor-D-homopregnanes derived from hecogenin.



¹ J. Fried, O. Wintersteiner, M. Moore, B. M. Iselin and A. Klingsberg, J. Am. Chem. Soc. **73**, 2970 (1951).

² K. Tsuda, S. Okuda and H. Kataoka, 14th Annual Meeting of Pharmaceutical Society of Japan (Sapporo, July 19, 1961).

³ J. Fried and A. Klingsberg, J. Am. Chem. Soc. **75**, 4929 (1953).

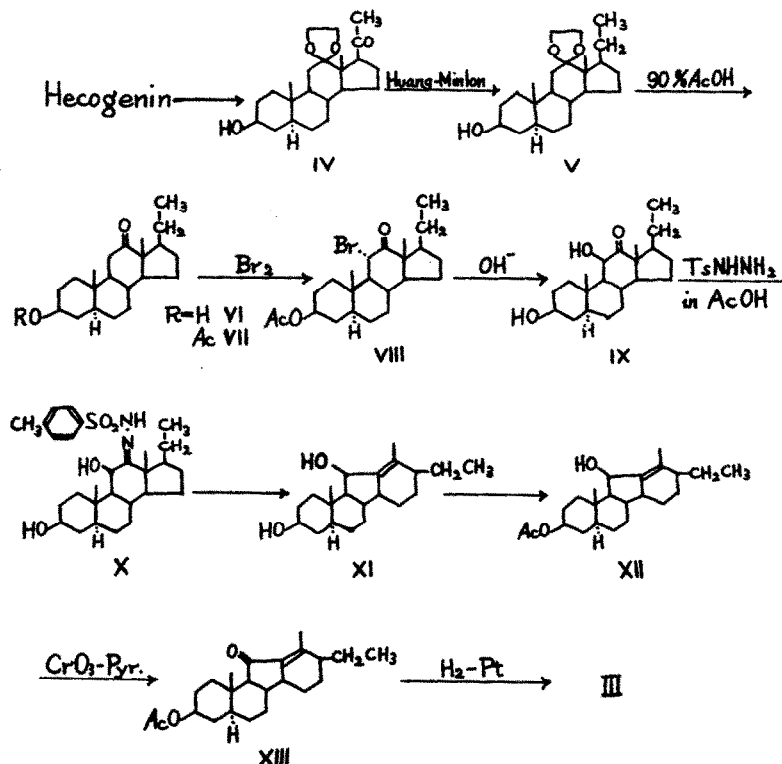
3 β -Hydroxy-12,12-ethylenedioxy-5 α -pregnan-20-one (IV)⁴ prepared in nine steps from hecogenin was reduced to 3 β -hydroxy-12,12-ethylenedioxy-5 α -pregnane (V) (m.p. 165°) by Huang-Minlon's method in 87% yield. Removal of the ethylenedioxy group by heating in 90% acetic acid gave 3 β -hydroxy-5 α -pregnane-12-one (VI) (m.p. 186°, 85% yield), which was converted into the acetate (VII) (m.p. 140°, 90% yield). Bromination of VII in benzene readily afforded 3 β -acetoxy-11 α -bromo-5 α -pregnane-12-one (VIII) (m.p. 148°, 50% yield), which was carefully hydrolyzed to 3 β ,11 β -dihydroxy-5 α -pregnane-12-one (IX) (m.p. 192-193°, 50% yield). The *p*-toluenesulfonylhydrazone of IX, (X) (m.p. 235° decomp.), which was obtained on treatment of IX with *p*-toluenesulfonylhydrazide in acetic acid, was heated with sodium ethyleneglycolate at 170°,⁵ to give XI, m.p. 190°, $[\alpha]_D^{25} +34.0^\circ$ (C, 0.94 CHCl₃), Found: C, 79.25; H, 10.85. Compound XI possesses a double bond exocyclic to a five-membered ring because of its relatively strong absorption at 1678 cm⁻¹ and its reduction with PtO₂ or Pd-C to a saturated mono-ol with loss of the 11-hydroxyl group suggesting the presence of an allylic structure. Treatment of XI with acetic anhydride-pyridine afforded a monoacetate (XII), m.p. 152°, $[\alpha]_D^{25} +31.2^\circ$ (C, 1.09 CHCl₃), IR: 3580 (11-hydroxyl), 1738 (3-acetyl), 1678 cm⁻¹ (double bond), Found: C, 76.50; H, 10.07, in 65% yield. On mild oxidation with CrO₃ in pyridine the acetate XII produced an α,β -unsaturated ketone (XIII), m.p. 143.5°, $[\alpha]_D^{25} -34.2^\circ$ (C, 1.35 CHCl₃), Found: C, 77.01; H, 9.55, in 70% yield.

⁴ D.N. Kirk, D.K. Patel and V. Petrow, *J. Chem. Soc.* 1957, 1046.

⁵ R. Hirschman, C.S. Snoddy, Jr., C.F. Hiskey and N.L. Wendler, *J. Am. Chem. Soc.* 75, 4929 (1953).

This substance has the same chromophore as jervine showing a UV maximum at 255 m μ (log ϵ =4.1), 354 m μ (log ϵ =1.89) and an IR spectrum closely resembling that of jervine in the carbonyl region.

Catalytic hydrogenation of XIII with PtO₂ gave a dihydro product, m.p. 118.2-118.9°, [α]_D-15.5° (C, 1.55 CHCl₃), Found: C, 76.95; H, 10.03, which was proved to be identical with III prepared from jervine³ by mixed melting point⁶ and comparison of IR spectra.



⁶ J. Fried and A. Klingsberg reported the melting point and specific rotation of III as 114-116°, [α]_D-8.6°(CHCl₃), but present sample from jervine shows m.p. 118-119° and [α]_D-15.8°(CHCl₃).

Since jervine and veratramine have been correlated with each other,⁷ the C-nor-D-homo ring system of both compounds has been confirmed conclusively. Moreover, the stereochemistry at 8, 9 and 14 has been shown to be that of conventional steroids.

Acknowledgement: The authors wish to express their thanks to Drs. K. Tsuda, S. Okuda, J. Fried and N. Masamune for their kind donation of III.

⁷ O. Wintersteiner and M. Moore, J. Am. Chem. Soc. 75, 4938 (1953).
Ch. Tamm and O. Wintersteiner, J. Am. Chem. Soc. 74, 3842 (1952).
O. Wintersteiner and N. Kosansky, J. Am. Chem. Soc. 74, 4474 (1952).